

5 α -Carboxystrictosidine†

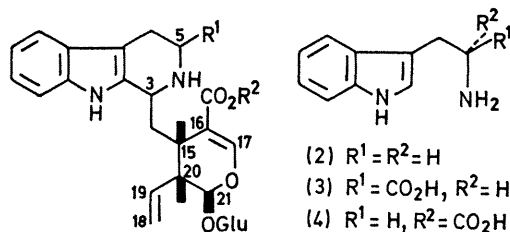
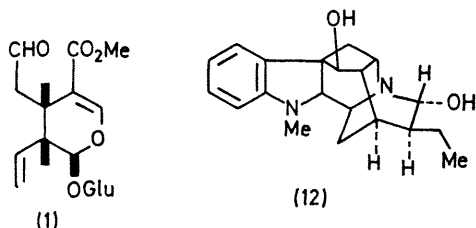
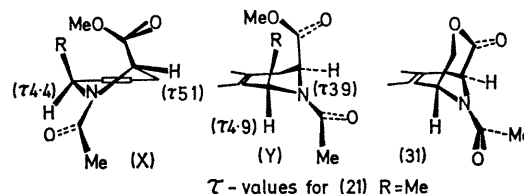
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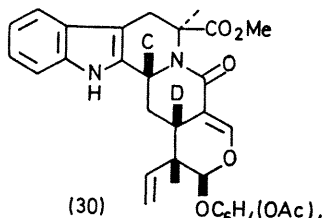
Summary The isolation, structural elucidation, biosynthesis, and partial synthesis of 5 α -carboxystrictosidine and related compounds are described.

CURRENT views on the biosynthesis of indole alkaloids imply that the first nitrogenous intermediate is produced by condensation of secologanin (1) with tryptamine (2).¹ However, van Tamelen² speculated on the possible significance of the corresponding tryptophan analogues and recently demonstrated their potential importance by the biogenetically modelled synthesis of ajmaline (12).³ Furthermore, the isolation of cordifoline,⁴ a 10-hydroxy- β -carboline analogue, demonstrates that at least in *Adina cordifolia* such precursors are likely. Accordingly our studies of *Rhazya* species were extended in a search for

been obtained completely homogeneous. The structures (7) and (9) are however consistent with the properties of the other glycosides which were characterised by conversion into strictosidine (5)⁵ and 5 α -methoxycarbonylstrictosidine (11) respectively.



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|------|-------------------------------------|---------------------|-------------------------|
| (5) | R ¹ = H | R ² = Me | 3 α |
| (6) | R ¹ = H | R ² = Me | 3 β |
| (7) | R ¹ = H | R ² = H | 3 α |
| (8) | R ¹ = CO ₂ H | R ² = Me | 3 α , 5 α |
| (9) | R ¹ = CO ₂ H | R ² = H | 3 α , 5 α |
| (10) | R ¹ = CO ₂ H | R ² = Me | 3 β , 5 α |
| (11) | R ¹ = CO ₂ Me | R ² = Me | 3 α , 5 α |
| (27) | R ¹ = CO ₂ H | R ² = Me | 3 α , 5 β |
| (26) | R ¹ = CO ₂ H | R ² = Me | 3 β , 5 β |



amino-acid glycosides. We now report the presence of substantial amounts of three such intermediates (each > 0.2% dry weight) in *R. orientalis* root extracts, and summarise the investigations which have led to the elucidation of their complete structures.

The amorphous nature of the new amino-acid glycosides complicates their characterisation and only 5 α -carboxystrictosidine (8) m.p. 232° [α]_D -280° (MeOH) has so far

Esterification of (7), (8), and (9) with diazomethane or diazoethane afforded their respective esters and *N*-alkyl esters, whose comparative study, together with their 18,19-dihydro-derivatives and acetates, by mass, i.r., u.v., and n.m.r. spectroscopy⁵ fully confirmed the structures. The main problem was then the determination of the stereochemistry of 5 α -carboxystrictosidine (8).

Initially it was necessary to confirm the expected 5 α -configuration and this was readily achieved by the excellent incorporation (10%) of [³H]-generally labelled-(*L*)-tryptophan using *R. orientalis*. Similarly support for the predicted secologanin stereochemistry (1) of the monoterpenoid glucoside moiety was obtained by feeding [*O*-methyl-³H] loganin (0.3% incorporation).

The next stage required the determination of the C-3-stereochemistry and here use was made of the established 5 α -configuration. Methoxide-catalysed epimerisation† of 5 α -methoxycarbonylstrictosidine penta-acetate (13), m.p. 200°, gave a separable mixture of monoacetates (14) and (15), which were shown to equilibrate under the reaction conditions. The predominant isomer (15), afforded a new penta-acetate (16) on acetylation. Similar treatment of 5 α -dihydromethoxycarbonylstrictosidine penta-acetate (17) gave an even greater preponderance of the 5 β -derivative (18). Examination of this series of C-5-epimers by n.m.r. spectroscopy revealed the presence of a low-field proton signal τ 3.9–4.1 (C₆D₆) in the 5 α -series only.

In order to study this distinguishing feature, methyl esters (19), m.p. 68°, [α]_D -92 (MeOH), and (20), m.p. 153°, [α]_D +15° (MeOH) were prepared by condensation of *L*-tryptophan (3) with acetaldehyde⁶ followed by esterification with methanolic hydrogen chloride. Acid-catalysed equilibrations demonstrated that the major isomer (19) was the thermodynamically more stable, suggesting its 3 α ,5 α stereochemistry and in agreement, the n.m.r. spectrum revealed a characteristic ABX pattern for the 5-H-6-H (*J*_{AB} 15, *J*_{AX} 4.5, *J*_{BX} 10.8 Hz). The n.m.r. spectrum of the minor isomer (20) further supported the stereochemical assignment.

† For indole alkaloid numbering see preceding communication.

‡ All methoxycarbonylamides subjected to C-5 equilibration are in the Table.

On acetylation (**19**) produced the amide (**21**), m.p. 195°, $[\alpha]_D +192^\circ$ (MeOH) whose n.m.r. spectrum possessed the characteristic low-field proton signal [τ 3.9 (CDCl₃)] which was absent from the epimeric amide (**22**) m.p. 209° $[\alpha]_D +13^\circ$ (MeOH). Owing to restricted rotation about the NCOCH₃ linkage each spectrum was composed of contributions from the two possible rotamers which aided their interpretation. Double irradiations strongly supported the 3 α ,5 α assignment for the major amide (**21**) which was interpretable in terms of rotamers (**21X**) and (**21Y**).

In contrast to the behaviour of 5-methoxycarbonylstrictosidine derivatives, on methoxide-catalysed equilibration the enantiomers (**21**) and (**23**), possessing the lowfield

terms of rotamer population and τ values for 3-H and 5-H.

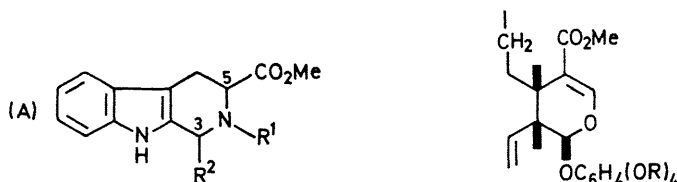
All the evidence now favoured the assignment of the 3 α -configuration to 5 α -methoxycarbonylstrictosidine pentaacetate (**13**). Final proof was obtained when secologanin (**1**),^{7,10,11} (fresh *R. orientalis* leaves) became available. Condensation of (**1**) with L-tryptophan (**3**) gave predominantly the *cis*-isomer 5 α -carboxystrictosidine (**8**) and 5 α -carboxyvincoside (**10**). A similar condensation starting with D-tryptophan (**4**) yielded the *cis*-isomer, 5 β -carboxyvincoside (**26**), and 5 β -carboxystrictosidine (**27**). Methylation followed by acetylation produced the corresponding penta-acetates which were interrelated by methoxide-catalysed equilibrations (**13**) \rightleftharpoons (**16**), (**28**) \rightleftharpoons (**29**).

Methoxycarbonyl-amides subjected to methoxide catalysed C-5 equilibrations

Structure (A) R ¹ = Ac R ² =	3 α ,5 α			3 α ,5 α			3 β ,5 α			3 β ,5 β		
	(I)	(II)	(III)	(I)	(II)	(III)	(I)	(II)	(III)	(I)	(II)	(III)
I	(13)	-50	32	(16)	-40	68	(29)	-86	55	(28)	-178	45
G	(14)	-25	32	(15)	-16	68						
J	(17)	-51	25	(18)	-62	75						
Me	(21)	+192	98	(24)	-14	2	(22)	+13	3	(23)	-191	97
Bu ^t		+28	30		+45	70		-47	70		-27	30
PhCH ₂		+107	25		-41	75		+40	75		-103	25
Pr ^t			2			98		^a			^a	

(I) Compound No.; (II) $[\alpha]_D^{25}$ (MeOH) except I, G, and J. (CHCl₃); (III) % on C-5 equilibration.

^a Prepared from (DL)tryptophan.



(19) R ¹ = H, R ² = Me	3 α ,5 α	G, R = H
(20) R ¹ = H, R ² = Me	3 β ,5 α	I, R = Ac
(25) R ¹ = Ac, R ² = CH ₂ OAc	3 α ,5 α	J, R = Ac vinyl reduced

signal, proved to be much more stable than their *trans*-isomers (**22**) and (**24**). This was attributed to the greater steric requirements encountered in rotamers (**X**) and (**Y**) when the methyl group was further substituted. Accordingly a range of such substituents was studied (Table).

In each case esterification of the condensation products gave a preponderance of the *cis*-isomer possessing the characteristic ABX pattern. Acetylation produced the corresponding amides whose n.m.r. spectra and thermodynamic stabilities substantiated the steric rationalisation. The *cis*-stereochemistry of the major condensation product was finally proved by utilising glycolic aldehyde and L-tryptophan which afforded the lactone (**31**), m.p. 232°, $[\alpha]_D +70^\circ$ (MeOH) convertible into acetoxy-ester (**25**), m.p. 124°, whose n.m.r. spectrum paralleled that of (**21**) in

5 α -Carboxyvincoside (**10**) undergoes simple lactamisation yielding, after methylation and acetylation, 5 α -methoxycarbonylvincoside lactam tetra-acetate (**30**) [5-H, τ 4.0 (CDCl₃)]. In conformational terms this observation is in complete agreement with the predicted stabilities of the c/d ring systems, for only in the 5 α -carboxyvincoside series are all the steric interactions, dipoles, and orbital-overlap requirements of the alkoxyacrylic lactam function, favourable. This rationalisation when applied to the ease of lactamisation¹ in the corresponding tryptamine-derived compounds strictosidine (**5**) and vincoside (**6**) offers a further vindication of the previous chemical correlation^{8,9} which proposed the 3 α -configuration for strictosidine (**5**).

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